

Femtosecond laser for surface structuring of new electrode materials

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Our New Energy World

Hydrogen generation and energy storage



Needs efficient generation of H_2

→ Electrolysis

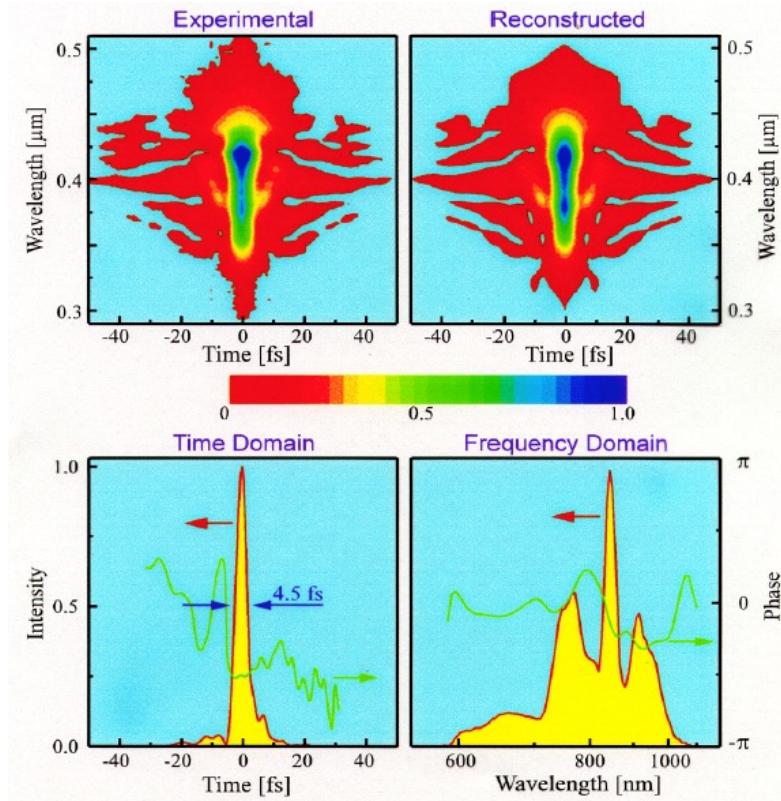
→ Optimization of electrodes



Rechargeable Zn-Air batteries

→ New electrode materials

Our working tool – femtosecond lasers pulses



1 femtosecond = 10^{-15} s

= 0,0000000000000001 s

... with amazing characteristics ...

Femtosecond Laser Surface Structures

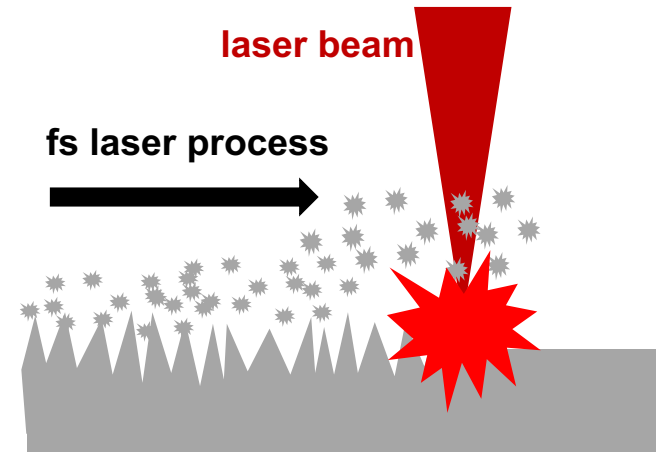
Structure Development

Number of laser pulses per spot on the surface^[1]:

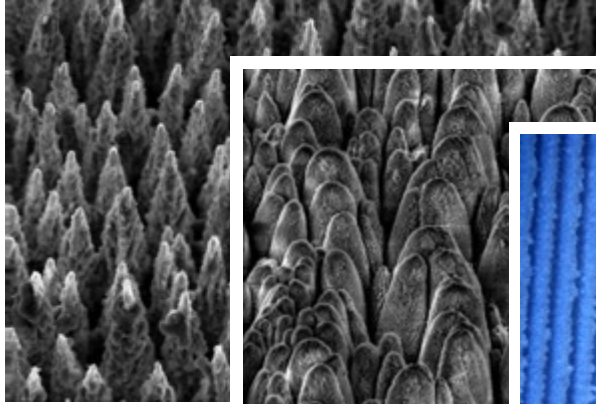
$N = 1 - 10 \rightarrow 10 - 25 \rightarrow 25 - 50 \rightarrow 50 - 100 \rightarrow \mathbf{100 - 250} \rightarrow \gg 1000$

LIPSS $\rightarrow$ ripples $\rightarrow$ grooves $\rightarrow$ cones $\rightarrow$ cones (10 μm) $\rightarrow$?

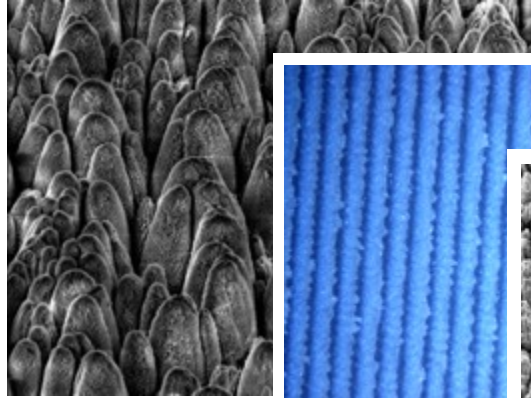
surface enlargement



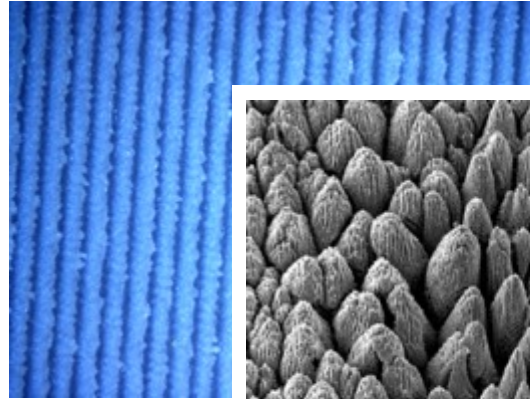
SurfaceLab – Femtosecond Laser Surface Processing



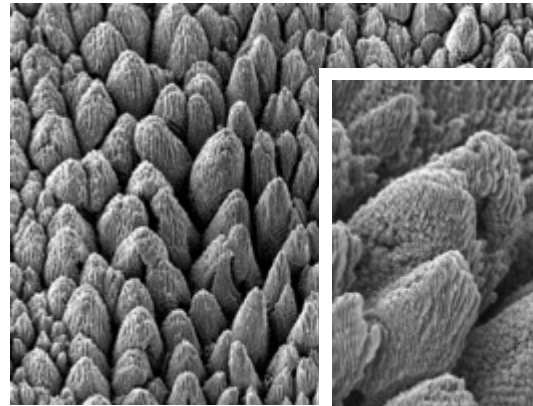
silicon



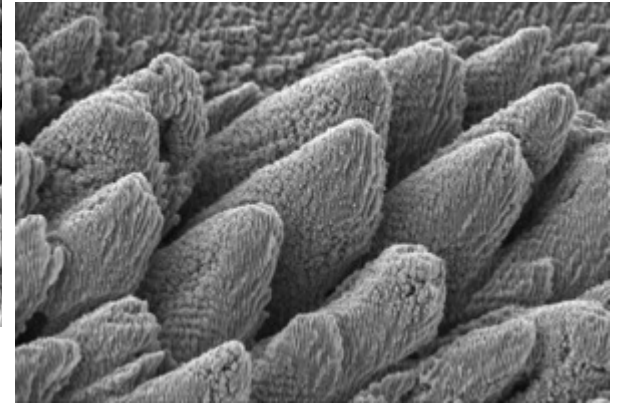
aluminium



polymer



titanium



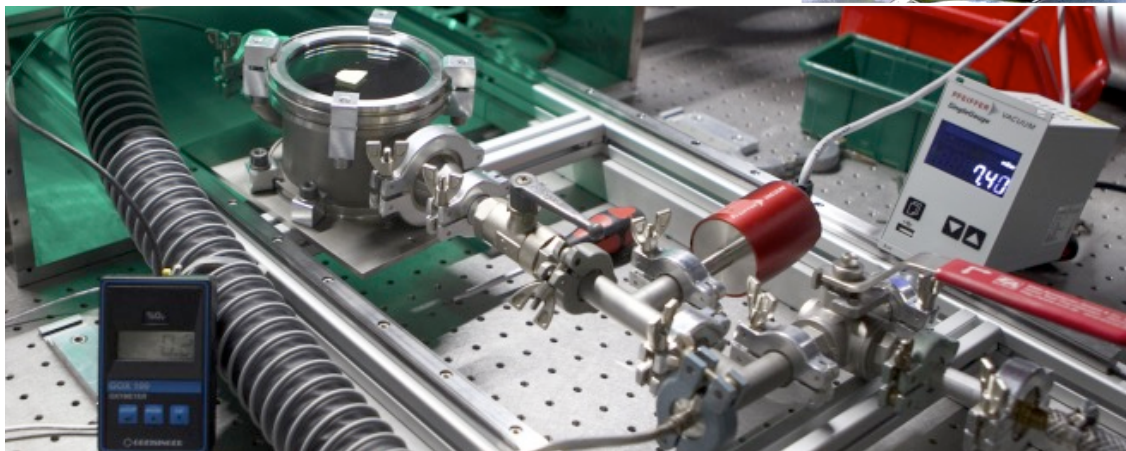
nickel

... zinc, silver, ceramic, copper, graphite, ...

Femtosecond Laser Surface Structures

Controlling Surface Chemistry

- To control wettability
- Optimized stoichiometry for catalysts
- Ar, N₂, CO₂, O₂, Cl₂, Br₂, I₂,



Pure metal

@ Argon:
2.5 atom-% O

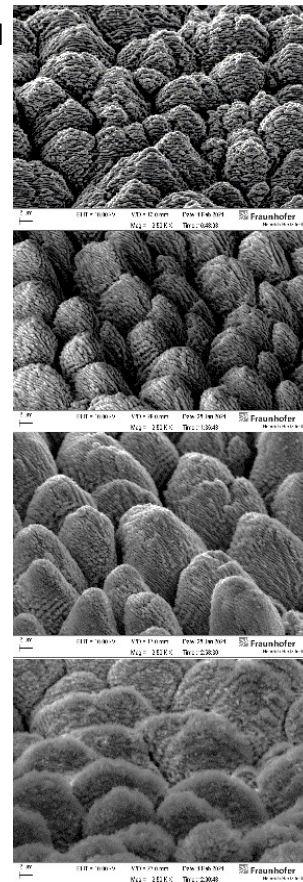
@ CO₂:
12 atom-% O

@ Air:
36 atom-% O

@ O₂:
46 atom-% O

Metal oxide

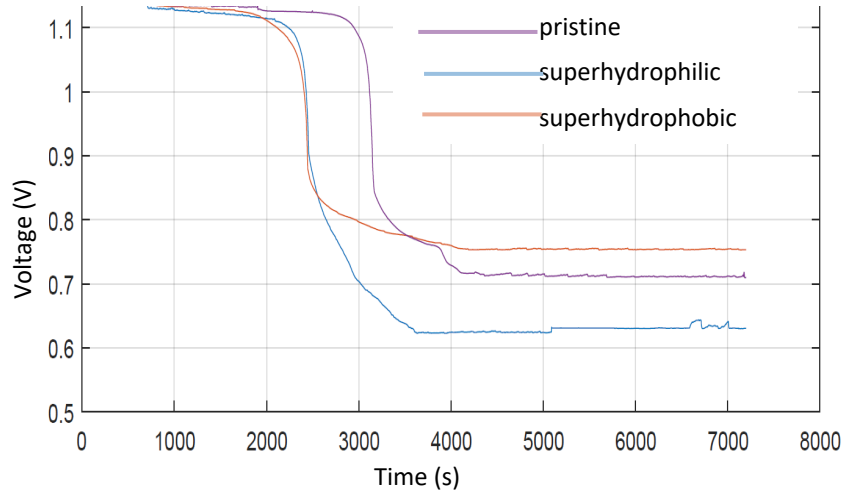
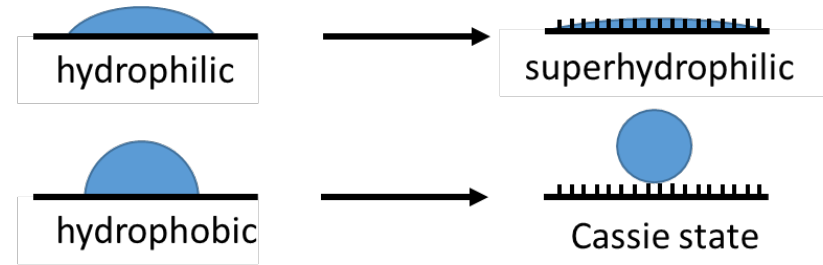
steel



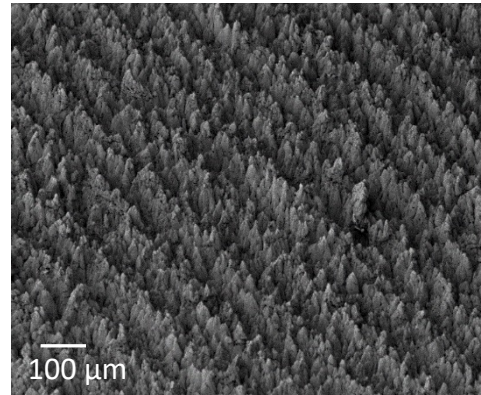
Femtosecond Laser Surface Structures

Controlling Wettability

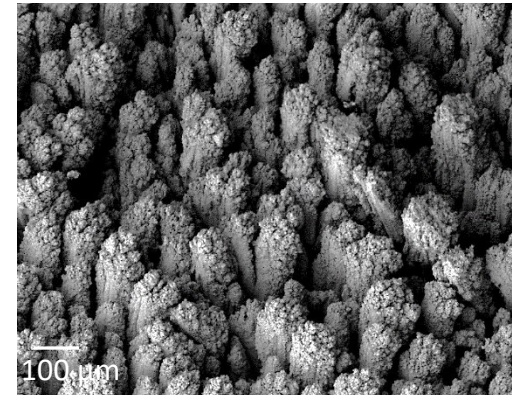
- Rechargeable zinc-air battery
- GDE water contact angle: 0° vs. 130° (pristine) vs. 180°
- Improved oxygen consumption



➤ Cell voltage - 0.1 V (0°) $\rightarrow$ + 0.05 V (180°)



$\Theta \rightarrow 0^\circ$

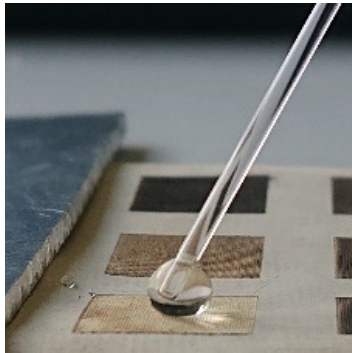


$\Theta \rightarrow 180^\circ$

Femtosecond Laser Surface Structures

Zinc-air Batteries

- Rechargeable zinc-air battery
- Key part: (bifunctional) gas diffusion electrode (GDE)
 - Ag/AgO (Co_3O_4) and hydrophobic binder
- Femtosecond laser structuring to optimize surface structure and wettability

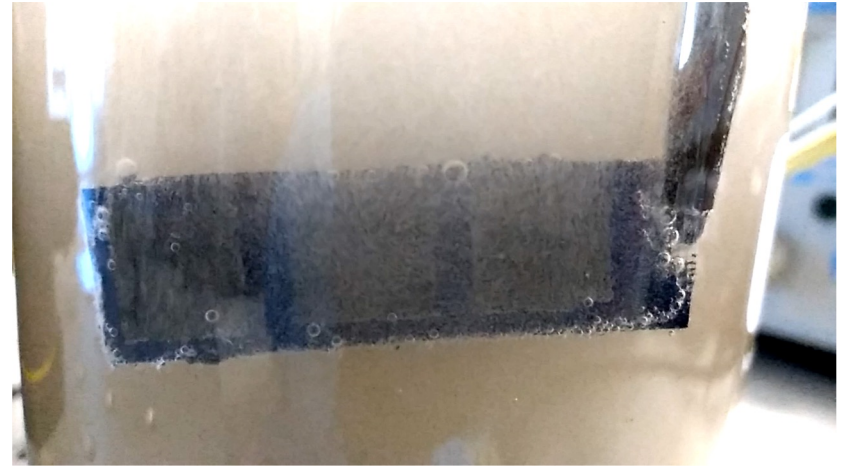


$\Theta \rightarrow 180^\circ$



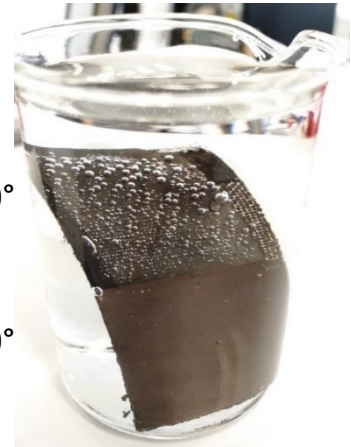
$\Theta \rightarrow 0^\circ$

Video: Oxygen evolution



$\Theta \rightarrow 180^\circ$

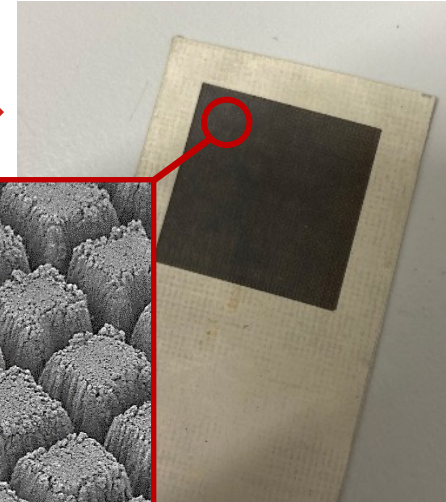
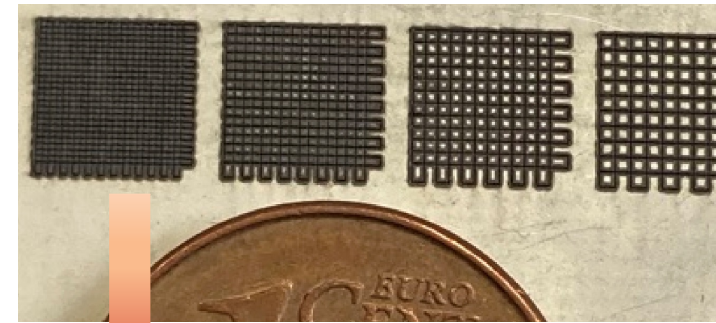
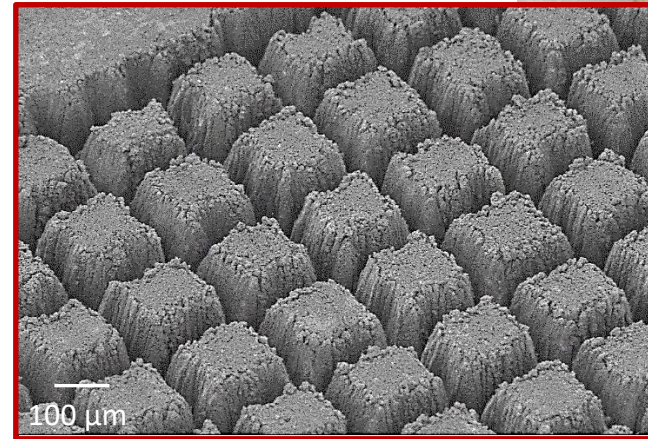
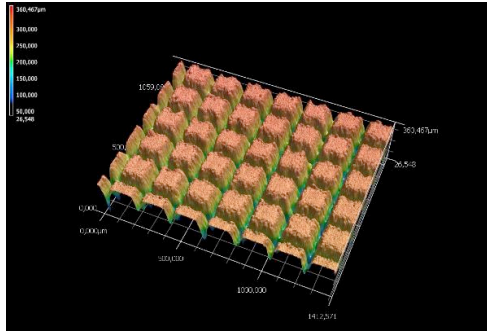
$\Theta \rightarrow 0^\circ$



Femtosecond Laser Surface Structures

Zinc-air-Batteries

- Bifunctional electrode must be optimized for oxygen consumption and evolution
- Porous electrode → significant contact area inside
 - Grid-like macroscopic surface structuring
 - 40 μm focus spot ($1/e^2$), 100 μm grid
 - Optimized mass transport phenomena



Femtosecond Laser Surface Structures

Hydrogen: Electrodes for Alkaline Water Electrolysis (AEL)

GEFÖRDERT VOM



Gefördert durch:



Niedersächsisches Ministerium
für Wissenschaft und Kultur

- Key Factor: Overpotential reduction

➤ Recent reviews define major aspects^[1]:

- Specific surface area (Tafel equation: overpotential η (V) $\sim$ current density i (A/cm²))
- Presence of gas bubbles on the surface of the electrode
- Electrocatalytic materials

• Surface enlargement

➤ Porosity

N pulses >1000

Superhydrophilicity

N pulses >250
processing at oxygen/air

Nickel oxides

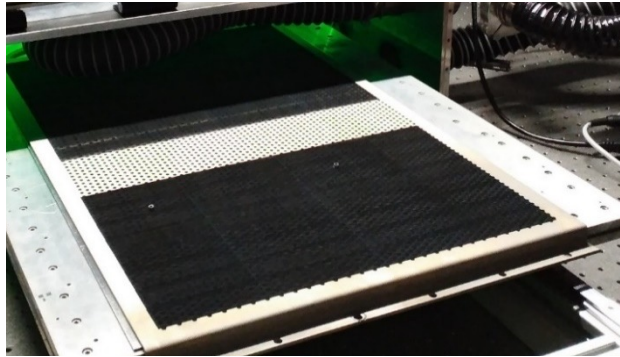
processing at oxygen/air



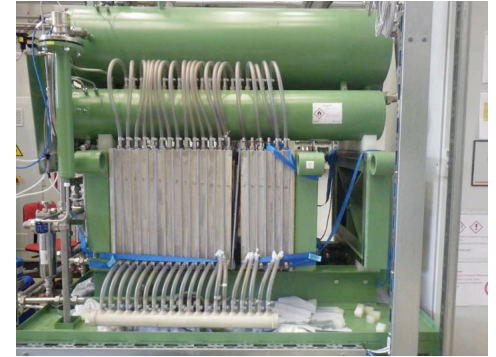
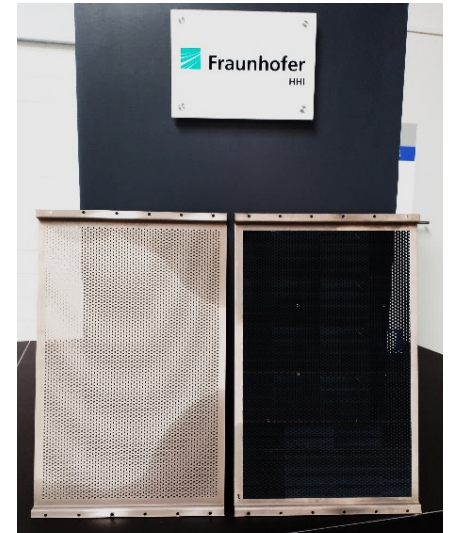
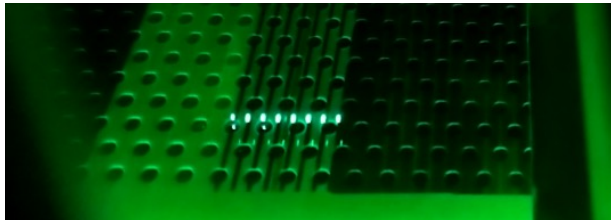
Femtosecond Laser Surface Structures

Large Scale Electrodes^[1]

- Base material: Steel coated with approx. 300 μm nickel (MTV)
- 0.66 x 0.41 m² structured on both sides (0.54 m² per electrode)
- N pulses $\sim 20 \times 2000$ ^[2]
- Processed at air
 - Porous microcones, nickel oxides, superhydrophilic surface



processing time: ~ 150 h / electrode
8-fold beam via diffractive element

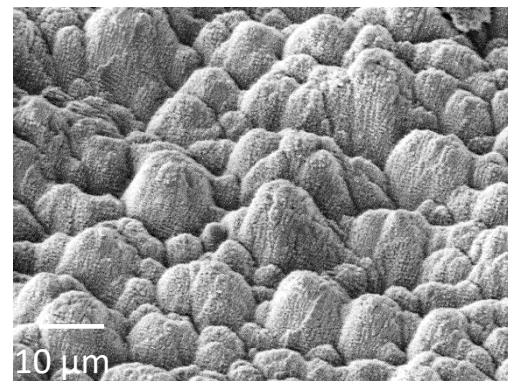
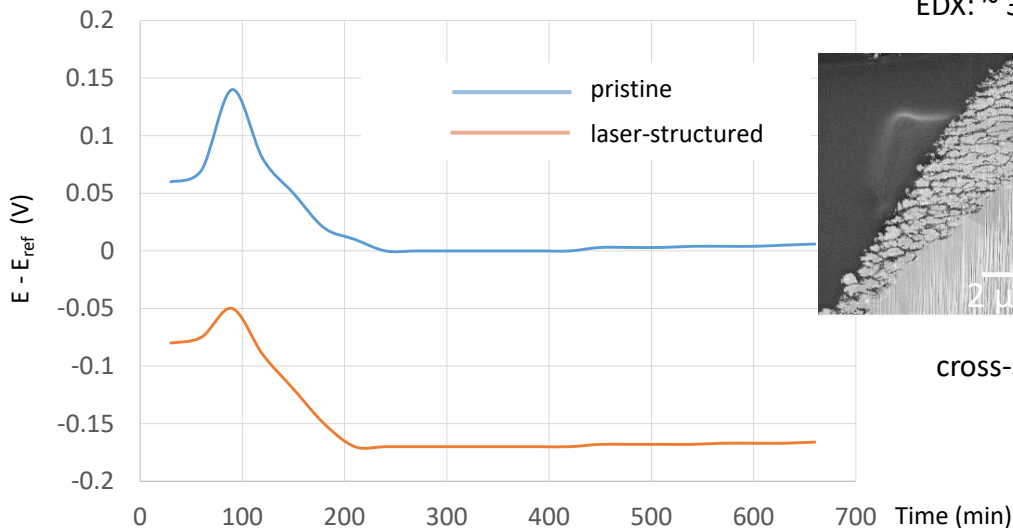


Femtosecond Laser Surface Structures

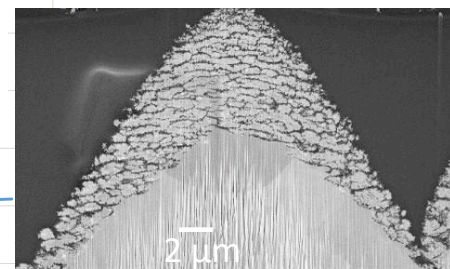
Electrodes for Alkaline Water Electrolysis (AEL)

- Surface enlargement $\sim 100\times$ (BET)
- Operation of electrodes at realistic conditions (Fraunhofer IFAM)
- 4 Electrodes laser-structured (both sides) vs. 4 pristine electrodes

- 750 A per electrode
- Overpotential reduced
- Process efficiency increased approx. 10 %



EDX: ~ 30 atom-% O

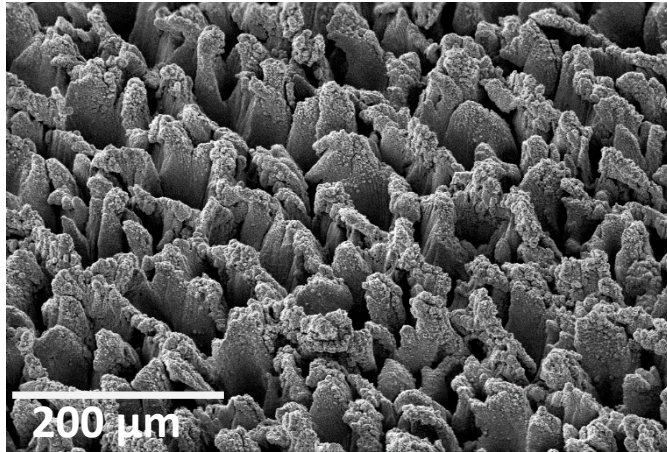


cross-section^[2]

Fem2Nano

Self-organized Structure Formation

- Microcones and related structural motifs
 - By physical and chemical ablation processes
 - Initiated by energy input (laser light) on the surface

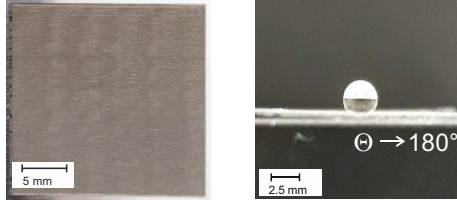


Fem2Nano

Transfer from Femtosecond to Nanosecond Laser Pulses

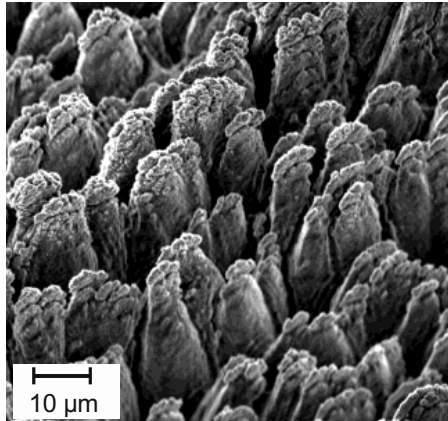
- Assisted by chemical ablation („etching“)

$\alpha = 77\%$



Process parameters:

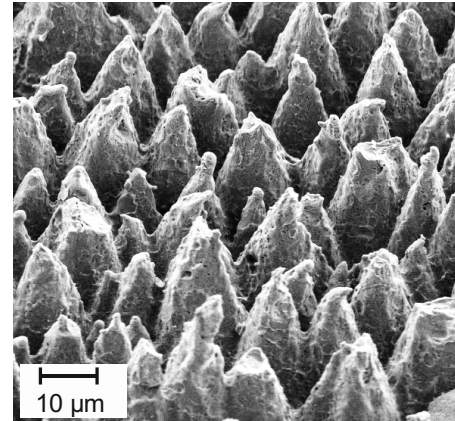
- $\tau = 60$ fs
- $J = 2.6$ J/cm²
- $N = 250$ pulses
- N₂-atmosphere



EDX [atom-%]:

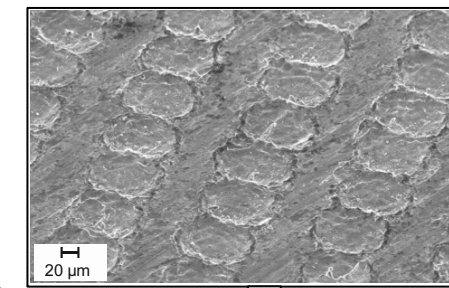
97 % Al, 3 % O

vs.



98 % Al, 2 % O

$\alpha = 56\%$



$J = 9.8$ J/cm²

$N = 512$ pulses

iodine/reduced pressure

Summary



Fraunhofer-Institut für Nachrichtentechnik
Heinrich-Hertz-Institut
HHI

Abteilung:
Faseroptische Sensorsysteme